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# 2,2',4,4',6,6'-Hexamethyl-4,4'-bi[4*H*-pyranyl]

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The title molecule,  $C_{16}H_{22}O_2$ , reveals  $C_i$  point symmetry in the crystal structure. The structure was disordered. The pyran ring is not planar; the O atom lies significantly out of the least-squares plane (ten times the r.m.s. deviation of all six atoms).

#### Comment

Bipyranyls and pyrylium salts are used for photochemical redox reactions. Bipyranyls may be applied as 2e donors in photoreductions, *e.g.* such as

 $2acrH^+ + (tmp)_2 + h\nu \rightarrow 2 tmp^+ + (acrH)_2$ 

where acrH<sup>+</sup> is acridinium and tmp<sup>+</sup> is 2,4,6-trimethylpyrylium. For photo-oxidations with pyrylium salts as e acceptors, attention has to be paid to the equilibrium of bipyranyl and the pyranyl radical,

 $tmp^+ + 2D + h\nu \rightarrow 2D^{+} + 2 tmp^- \leftarrow K \rightarrow (tmp)_2;$ 

where D is an e donor.

The title molecule, (I), reveals  $C_i$  point symmetry in the crystal structure. The structure was disordered in such a way



that each of three C atoms of the pyran ring occupies statisticially two sites with a ratio of 89/11. The sites of the other three ring atoms are not split. This means that the two ring positions are arranged like a roof with an interfacial angle of about  $24^{\circ}$ . A second crystal studied showed the same effect, with a similar occupation ratio of 81/19 and a corresponding interfacial angle of about  $27^{\circ}$ . The pyran ring is not planar; the O atom lies significantly out of the least-squares plane (ten times the r.m.s. deviation of all six atoms).

#### **Experimental**

The title compound was synthesized according to Balaban *et al.* (1964) and crystallized from ethanol.

Crystal data  $D_x = 1.132 \text{ Mg m}^{-3}$  $C_{16}H_{22}O_2$  $M_r = 246.34$ Mo  $K\alpha$  radiation Monoclinic,  $P2_1/c$ Cell parameters from 5000 a = 7.842 (2) Åreflections b = 11.458(3) Å  $\theta = 2-24^{\circ}$  $\mu = 0.073 \text{ mm}^{-1}$ c = 8.824 (2) A  $\beta = 114.26 \ (3)^{\circ}$ T = 180 (2) K $V = 722.9 (3) \text{ Å}^3$ Plate, colourless Z = 2 $0.52\,\times\,0.40\,\times\,0.22$  mm

#### Data collection

Stoe IPDS diffractometer  $\varphi$ -rotation,  $\varphi$ -incr. = 1.5°, 160 exposure scans 5901 measured reflections 1716 independent reflections 1393 reflections with  $I > 2\sigma(I)$ 

#### Refinement

refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.044$   $wR(F^2) = 0.119$  S = 1.0391716 reflections 121 parameters H atoms treated by a mixture of independent and constrained  $R_{int} = 0.033$   $\theta_{max} = 28.07^{\circ}$   $h = -10 \rightarrow 10$   $k = -15 \rightarrow 15$   $l = -11 \rightarrow 11$ Intensity decay: none

$$\begin{split} &w = 1/[\sigma^2(F_o^2) + (0.0650P)^2 \\ &+ 0.1273P] \\ &where \ P = (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\rm max} = 0.003 \\ \Delta\rho_{\rm max} = 0.22 \ {\rm e} \ {\rm \AA}^{-3} \\ \Delta\rho_{\rm min} = -0.16 \ {\rm e} \ {\rm \AA}^{-3} \end{split}$$

The structure is disordered so that split atomic positions of C3, C4 and C5 had to be introduced for the least squares refinement. Without such a 'split model', all parameters of quality for a structure determination become worse as follows:  $R_1(gt)$  increases from 0.0438 to 0.0751;  $wR_2(all)$  increases from 0.1185 to 0.2182; GoF changes from 1.039 to 1.066; the long axes of the displacement ellipsoids of C3, C4 and C5 increase by a factor of about 1.35; and last, but not least, the three highest peaks in the final difference map become significantly larger (1.02, 0.82 and 0.77 e Å<sup>-3</sup>) than its r.m.s. (0.07 e Å<sup>-3</sup>); the peaks lie nearest to C4 (1.09 Å), C5 (0.82 Å) and C3 (0.79 Å), respectively.

Data collection: *IPDS*-2.87 (Stoe & Cie, 1997); cell refinement: *IPDS*-2.87; data reduction: *IPDS*-2.87; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1990); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); software used to prepare material for publication: *SHELXL*97.

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